

## **Technical Manual**

# **ColorFluor Glucose Oxidase Activity Assay Kit**

**Catalogue Code: BA0116**

**Pack Size: 100 assays**

**Research Use Only**



4. Read optical density immediately ( $OD_0$ ) at 570 nm (550-585 nm). Incubate 20 min at room temperature, and then read optical density again ( $OD_{20}$ ).

#### FLUORIMETRIC PROCEDURE

For fluorimetric assays, the linear detection range is 0.002 to 1.5 U/L glucose oxidase. Dilute the standards from *Colorimetric Procedure* 10× with  $dH_2O$  to obtain standards at 20, 12, 6 and 0  $\mu M$   $H_2O_2$ .

Transfer 20  $\mu L$  standards and 20  $\mu L$  samples into separate wells of a *black* 96-well plate.

Add 80  $\mu L$  Working Reagent (see *Colorimetric Procedure*), tap plate to mix.

Read fluorescence immediately ( $F_0$ ) at  $\lambda_{ex/em} = 530/585$  nm, incubate 20 min at room temperature, and then read fluorescence again ( $F_{20}$ ).

#### CALCULATION

Subtract blank  $OD_{20}$  or  $F_{20}$  (water, #4) from all standard  $OD_{20}$  or  $F_{20}$  values and plot the  $\Delta OD$  or  $\Delta F$  against standard concentrations. Determine the slope using linear regression. Calculate the  $\Delta OD_{Sample}$  or  $\Delta F_{Sample}$  of all samples by subtracting  $OD_0$  or  $F_0$  from  $OD_{20}$  or  $F_{20}$  for each sample. Do the same for the blank (water, standard #4) to get  $\Delta OD_{Blank}$  or  $\Delta F_{Blank}$ . Calculate the activity using the equation below:

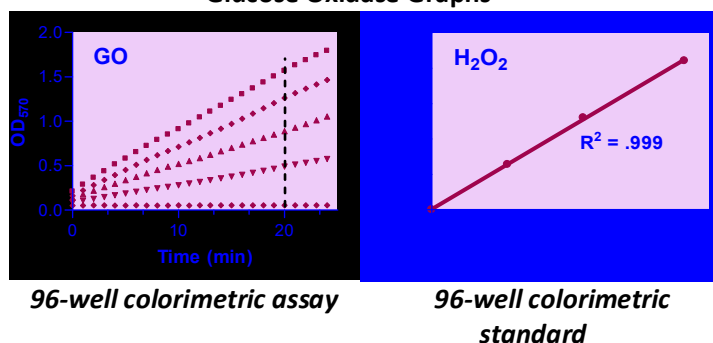
$$GO \text{ Activity} = \frac{\Delta R_{SAMPLE} - \Delta R_{BLANK}}{\text{Slope } (\mu M^{-1}) \cdot t} \times n \text{ (U/L)}$$

Where  $\Delta R_{Sample}$  and  $\Delta R_{Blank}$  are the change in optical density or fluorescent values of the sample and blank, respectively. *Slope* is the slope of the  $H_2O_2$  standard curve, *t* is the incubation time (20 minutes), and *n* is the dilution factor.

**Notes:** If the calculated sample glucose concentration is higher than 10 U/L in colorimetric assay or 1.5 U/L in fluorimetric assay, dilute sample in water and repeat the assay. Multiply result by the dilution factor (*n*). For samples with low Glucose Oxidase activity, the incubation time can be increased.

**Unit definition:** 1 U/L of Glucose Oxidase catalyzes 1  $\mu$ mole of  $H_2O_2$  per minute at pH 7.0 and room temperature.

#### Glucose Oxidase Graphs



#### MATERIALS REQUIRED, BUT NOT PROVIDED

Pipetting devices, centrifuge tubes, clear flat-bottom 96-well plates, black 96-well plates and plate reader.

#### LITERATURE

1. Raba J, Mottola HA. (1995). Glucose Oxidase as an Analytical Reagent. *Critical Reviews in Analytical Chemistry* 25(1):1-42.
2. Harris JM, Reyes C, Lopez GP. (2013). Common Causes of Glucose Oxidase Instability in in vivo Biosensing: a Brief Review. *J Diabetes Sci Technol* 7(4):1030-8.

3. Ferri S, Kojima K, Sode K. (2011). Review of glucose oxidases and glucose dehydrogenases: a bird's eye view of glucose sensing enzymes. J Diabetes Sci Technol 5(5):1068-76.

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