

Ammonia Colorimetric Assay Kit

(Catalog #BN00632; 100 assays; Store kit at -20° C)

I. Introduction:

Ammonia is an important source of nitrogen for living systems. Nitrogen is required for the synthesis of amino acids, which are the building blocks of protein. Ammonia is a metabolic product which is created through amino acid deamination. It plays an important role in both normal and abnormal animal physiology such as acid/base balance. Assay Genie provides a rapid, simple, sensitive, and reliable assay suitable for research and high throughput assay of Ammonia or Ammonium. In the assay, Ammonia or ammonium is converted to a product that reacts with the GenieRed probe to generate color (OD 570 nm) which can be easily quantified by plate reader. The kit can detect 1 nmol (~20 µM) of ammonia or ammonium, much more sensitive than measuring NADPH based ammonia assay.

II. Kit Contents:

| Components | 100 assays | Cap Color | Part Number |
|-------------------------------------|------------|-----------|-------------|
| Ammonia Assay Buffer | 25 ml | WM | BN00632-1 |
| GenieRed Probe in DMSO | 200 µl | Red | BN00632-2 |
| Enzyme Mix (lyophilized) | 1 vial | Green | BN00632-3 |
| Developer | 1 vial | Orange | BN00632-4 |
| Converting Enzyme (Lyophilized) | 1 vial | Blue | BN00632-5 |
| NH ₄ Cl Standard (10 mM) | 100 µl | Yellow | BN00632-6 |

III. Storage and Handling:

Store the kit at -20° C, protect from light. Allow Assay Buffer to warm to room temperature before use. Briefly centrifuge vials before opening. Read the entire protocol before performing the assay. **All the solution in this kit should be kept capped when not in use to prevent absorption of ammonia from the air.**

IV. Reagent preparation:

GenieRed Probe: Warm to room temperature before use. Store at -20°C, protect from light and moisture.

Enzyme Mix, Developer and Converting Enzyme: Dissolve in 220 µl Assay Buffer separately. Aliquot to prevent multiple freeze/thaw cycle. Store at -20°C. Use within two months.

V. Ammonia Assay Protocol:

1. Standard Curve Preparation:

Dilute the Ammonium Chloride standard solution to 1 mM by adding 10 µl of the 10 mM Ammonium Chloride Standard to 90 µl of ddH₂O, mix well. Add 0, 2, 4, 6, 8, 10 µl into each well individually. Adjust volume to 50 µl/well with Assay Buffer to generate 0, 2, 4, 6, 8, 10 nmol/well of the Ammonium Chloride Standard.

2. Sample Preparations:

Tissues (20 mg) or cells (2x10⁶) can be homogenized in 100 µl Assay Buffer, centrifuge at 13,000 xg for 10 minutes to remove insoluble material. Liquid sample can be tested directly. Add 2-50 µl sample to 96 well plate, bring the volume to 50 µl/well with Assay Buffer. For unknown samples, we suggest testing several different doses of samples to make sure the readings are within the standard curve range.

Note: Pyruvate in samples will interfere with the assay. If significant amount of pyruvate is suspected in your sample, set a Sample Control as in step 3. The pyruvate reading can be subtracted from sample readings.

3. **Reaction Mix:** Mix enough reagents for the number of assays to be performed. For each well, prepare a total 50 µl Reaction Mix.

| | Sample | Sample Control |
|----------------------|--------|----------------|
| Ammonia Assay Buffer | 42 µl | 44 µl |
| GenieRed Probe | 2 µl | 2 µl |
| Enzyme Mix | 2 µl | 2 µl |
| Developer | 2 µl | 2 µl |
| Converting Enzyme | 2 µl | 0 µl |

Add 50 µl of the **Reaction Mix** to each well containing the NH₄Cl Standard and test samples. Add 50 µl Sample Control Mix to Sample Control. Mix well. Incubate the reaction for 60 min at 37°C, protect from light.

4. **Measurement:** Measure O.D. 570 nm in a micro plate reader.

5. **Calculation:** Correct background by subtracting the value derived from the 0 NH₄Cl from all readings (The background reading can be significant and must be subtracted from readings). Subtract the Sample Control readings from sample readings. Plot NH₄Cl standard Curve, NH₄Cl concentrations of the samples can then be calculated:

$$C = S_a/S_v \text{ nmol/}\mu\text{l or mM,}$$

where S_a is the sample amount (in nmol) from standard curve.

S_v is the sample volume (µl) added into the wells.

NH₄⁺ Molecular Weight is 18.04 g/mol.

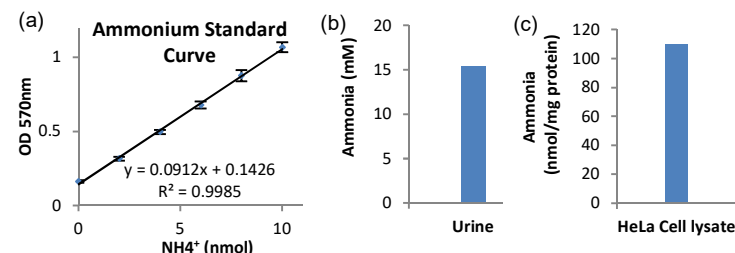


Figure. Ammonia Standard Curve (a). Ammonia concentration in human urine (25 µl, 100 times diluted) (b) and HeLa cell lysate (3 µl, 10.8 mg/ml) (c). Assay was performed according to the kit protocol.

FOR RESEARCH USE ONLY! Not to be used on humans.

GENERAL TROUBLESHOOTING GUIDE:



| Problems | Cause | Solution |
|---|---|--|
| Assay not working | <ul style="list-style-type: none"> • Use of ice-cold assay buffer • Omission of a step in the protocol • Plate read at incorrect wavelength • Use of a different 96-well plate | <ul style="list-style-type: none"> • Assay buffer must be at room temperature • Refer and follow the data sheet precisely • Check the wavelength in the data sheet and the filter settings of the instrument • Fluorescence: Black plates ; Luminescence: White plates ; Colorimeters: Clear plates |
| Samples with erratic readings | <ul style="list-style-type: none"> • Use of an incompatible sample type • Samples prepared in a different buffer • Samples were not deproteinized (if indicated in datasheet) • Cell/ tissue samples were not completely homogenized • Samples used after multiple free-thaw cycles • Presence of interfering substance in the sample • Use of old or inappropriately stored samples | <ul style="list-style-type: none"> • Refer data sheet for details about incompatible samples • Use the assay buffer provided in the kit or refer data sheet for instructions • Use the 10 kDa spin cut-off filter or PCA precipitation as indicated • Use Dounce homogenizer (increase the number of strokes); observe for lysis under microscope • Aliquot and freeze samples if needed to use multiple times • Troubleshoot if needed, deproteinize samples • Use fresh samples or store at correct temperatures till use |
| Lower/ Higher readings in Samples and Standards | <ul style="list-style-type: none"> • Improperly thawed components • Use of expired kit or improperly stored reagents • Allowing the reagents to sit for extended times on ice • Incorrect incubation times or temperatures • Incorrect volumes used | <ul style="list-style-type: none"> • Thaw all components completely and mix gently before use • Always check the expiry date and store the components appropriately • Always thaw and prepare fresh reaction mix before use • Refer datasheet & verify correct incubation times and temperatures • Use calibrated pipettes and aliquot correctly |
| Readings do not follow a linear pattern for Standard curve | <ul style="list-style-type: none"> • Use of partially thawed components • Pipetting errors in the standard • Pipetting errors in the reaction mix • Air bubbles formed in well • Standard stock is at an incorrect concentration • Calculation errors • Substituting reagents from older kits/ lots | <ul style="list-style-type: none"> • Thaw and resuspend all components before preparing the reaction mix • Avoid pipetting small volumes • Prepare a master reaction mix whenever possible • Pipette gently against the wall of the tubes • Always refer the dilutions in the data sheet • Recheck calculations after referring the data sheet • Use fresh components from the same kit |
| Unanticipated results | <ul style="list-style-type: none"> • Measured at incorrect wavelength • Samples contain interfering substances • Use of incompatible sample type • Sample readings above/below the linear range | <ul style="list-style-type: none"> • Check the equipment and the filter setting • Troubleshoot if it interferes with the kit • Refer data sheet to check if sample is compatible with the kit or optimization is needed • Concentrate/ Dilute sample so as to be in the linear range |
| Note: The most probable list of causes is under each problem section. Causes/ Solutions may overlap with other problems. | | |